Nucleophilic Micelles. II. The Effect on the Rate of Solvolysis of Neutral, Positively, and Negatively Charged Esters of Varied Chain Length when Incorporated into Nonfunctional and Functional Micelles of Neutral, Positive, and Negative Charge [*J. Am. Chem. Soc.*, **90**, 1333 (1968)]. By THOMAS C. BRUICE, J. KATZHENDLER, and LEO R. FEDOR, Department of Chemistry, University of California at Santa Barbara, Santa Barbara, California 93106.

Figures 1–5 were printed without identifying ordinates and abscissas. The corrected figures should be as shown below.

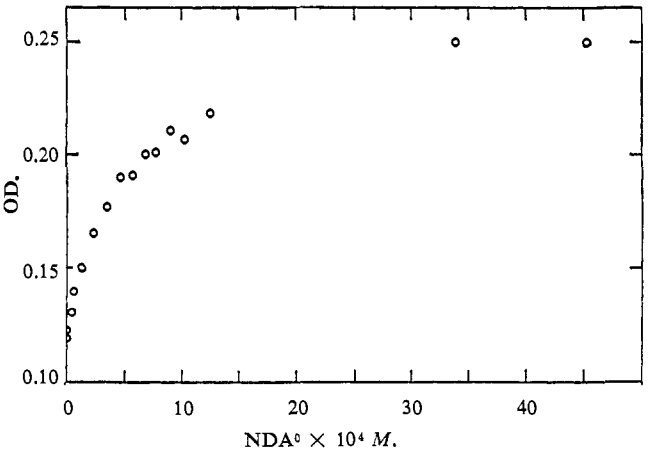

Figure 1. Absorbance of pinacyanol chloride [0.0027% (w/v); 610 mμ; μ = 0.5; pH 8.84] as a function of NDA⁰ concentration. The values of OD were found to be constant from [NDA⁰] = 3.4 × 10^{-3} to 9.6 × 10^{-3} *M* (last five points not shown).

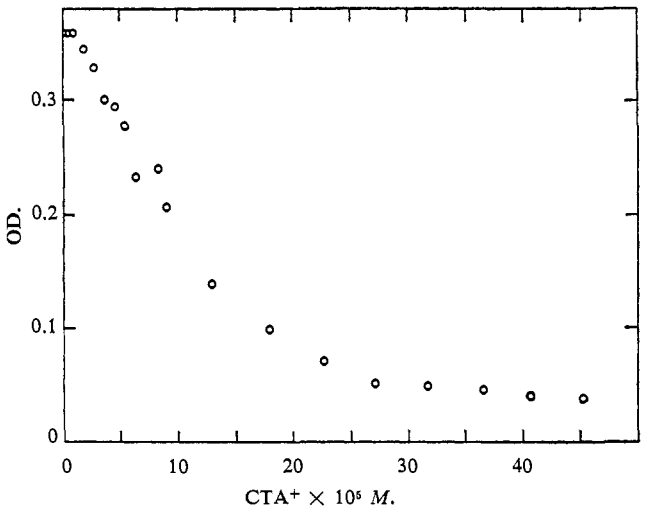

Figure 2. Absorbance of phenolphthalein [0.0022%(w/v); 560 mμ; μ = 0.5; pH 8.95] as a function of CTA^+ concentration.

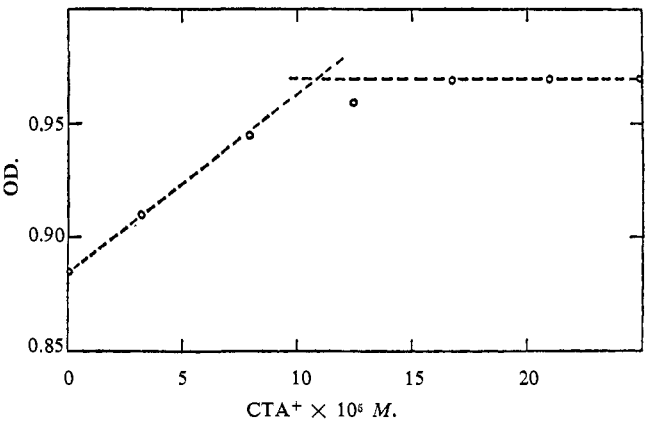

Figure 3. Dependence of the absorbance of NE_5^- on CTA^+ concentration (250 mμ; pH 6.17; μ = 0.5).

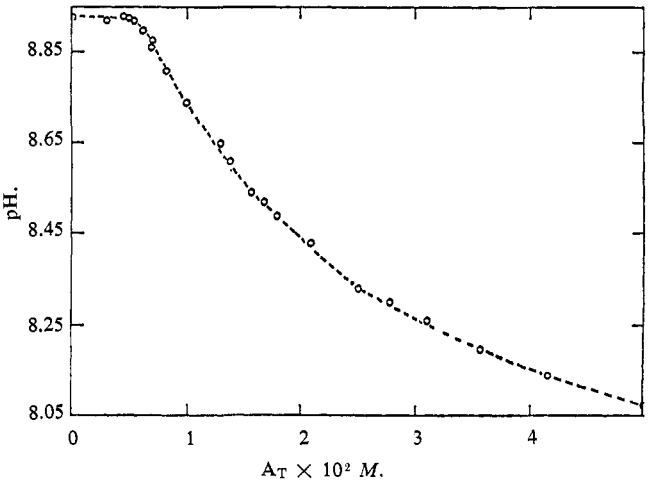

Figure 4. The pH of half-neutralized solutions of the amine hydrochloride of A_{10}^+ as a function of the concentration of A_T (= $[A_{10}^+]$ + $[A_{10}H^{2+}]$) at μ = 0.1.

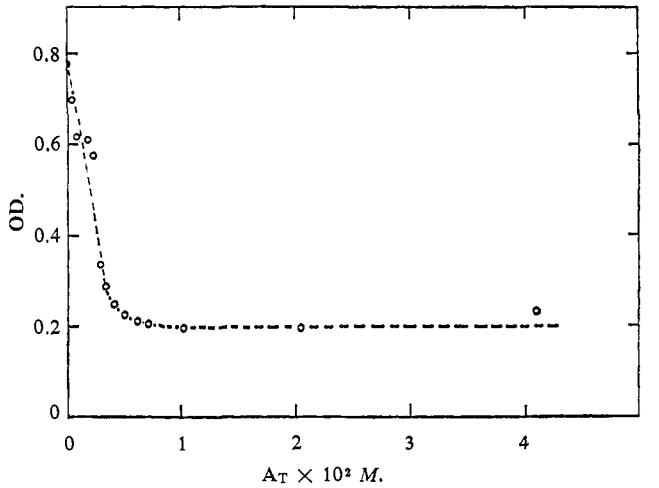

Figure 5. Absorbance of phenolphthalein [0.0045% (w/v); 560 mμ; μ = 0.1; pH 9.07] as a function of the concentration of A_T (= $[A_{10}^+]$ + $[A_{10}H^{2+}]$).